

You must copy each question.

Next to each question you must sketch the pattern of the trend that was presented in your notes.

1. Which one has the larger radius
 - a. F
 - b. Li
 - c. B
 - d. Be
2. Which one has the higher electronegativity
 - a. N
 - b. C
 - c. B
 - d. F
3. Which one has the higher ionization energy
 - a. Xe
 - b. Kr
 - c. Ar
 - d. Ne
4. Which one has the larger radius
 - a. Mg
 - b. Be
 - c. Ba
 - d. Ra
5. Which one has the higher electronegativity
 - a. I
 - b. Te
 - c. Sb
 - d. Cd
6. Which one has the higher ionization energy
 - a. Li
 - b. K
 - c. Rb
 - d. Cs
7. Which one has the larger radius
 - a. Li
 - b. K
 - c. Rb
 - d. Cs
8. Which one has the higher ionization energy
 - a. P
 - b. N
 - c. Sb
 - d. Bi