**Name\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ Period \_\_\_\_\_\_**

**Types of Chemical Reactions Lab**

**Station 1**

**Balanced equation:**

**\_\_\_Mg + \_\_\_O2 🡪 \_\_\_MgO**

**Reaction Type:**

**Observation list: Use bullet points**

**Station 2**

**Balanced equation:**

**\_\_\_CuSO4. 5 H2O 🡪 ­\_\_\_CuSO4 + \_\_\_H2O**

**Reaction Type:**

**Observation list: Use bullet points**

**Station 3**

**Balanced equation:**

**­\_\_Zn + \_\_HCl 🡪 \_\_ZnCl2 + \_\_H2**

**Reaction Type:**

**Observation list: Use bullet points**

**Station 4**

**Balanced equation:**

**\_\_\_KI + \_\_\_ Pb(NO3)2 🡪 \_\_\_KNO3 + \_\_\_PbI2**

**Reaction Type:**

**Observation list: Use bullet points**

**Station 5**

**Balanced equation:**

**\_\_\_NaOH + ­­\_\_\_HCl 🡪 \_\_\_NaCl + \_\_\_H2O**

**Reaction Type:**

**Observation list: Use bullet points**

**Station 6**

**Balanced equation:**

**\_\_\_C2H6O + \_\_\_O2 🡪 \_\_\_CO2 + \_\_\_H20**

**Reaction Type:**

**Observation list: Use bullet points**